

Physical Science

Grade 7

Written by Tracy Bellaire

About this Book

The experiments in this book fall under thirteen topics that relate to three aspects of physical science: **Structural Form and Function, Heat and Temperature, Chemistry of Pure Substances and Mixtures**. In each section you will find teacher notes designed to provide you guidance with the learning intention, the success criteria, materials needed, a lesson outline, as well as provide some insight on what results to expect when the experiments are conducted. Suggestions for differentiation are also included so that all students can be successful in the learning environment.



About the Author:

Tracy Bellaire is an experienced teacher who continues to be involved in various levels of education in her role as Differentiated Learning Resource Teacher in an elementary school in Ontario. She enjoys creating educational materials for all types of learners, and providing tools for teachers to further develop their skill set in the classroom. She hopes that these lessons help all to discover their love of science!

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At A Glance

Learning Intentions	Structural Forms	Forces on Structures	Form and Stability	Structure and Function	Under Construction	Design and Production	Temperature Change	Heat Transfer	Heat in Our Homes	Solutions	Mixtures	Mixing and Separating	Physical or Chemical?
Knowledge and Understanding Content													
• recognize and classify structural forms and the materials used in their construction	•												
• identify internal and external forces; describe the effects of internal and external forces on structures		•			•								
• recognize and describe the importance of symmetry and center of gravity on a structure when supporting a load			•		•								
• identify and describe supports used to strengthen structures and fasteners used to contribute to their function				•	•								
• investigate the impact of ergonomic design and other factors that determine product success in the market place					•								
• use the particle theory of matter to compare how heat affects the motion of particles in a solid, a liquid, and a gas						•				•			
• describe and investigate how heat is transferred through conduction, convection, and radiation							•						
• describe various sources of heat energy, and investigate how it is acquired, used, and conserved in our homes								•					
• identify the components of a solution, and describe the difference between saturated and unsaturated solutions										•			
• distinguish between pure substances, mechanical mixtures, emulsions, suspensions, and solutions											•		
• investigate the interactions of solids and liquids, and how a material can be recovered from a mixture or solution												•	
• differentiate between physical and chemical changes and determine their versatilities													•
Thinking Skills and Investigation Process													
• make predictions, formulate questions, and plan an investigation		•	•	•	•		•	•		•	•	•	•
• gather and record observations and findings using drawings, tables, written descriptions	•	•	•	•	•	•	•	•	•	•	•	•	•
• recognize and apply safety procedures in the classroom	•	•	•	•	•	•	•	•	•	•	•	•	•
Communication													
• communicate the procedure and conclusions of investigations using demonstrations, drawings, and oral or written descriptions, with use of science and technology vocabulary	•	•	•	•	•	•	•	•	•	•	•	•	•
Application of Knowledge and Skills to Society and the Environment													
• evaluate the importance to society, the economy, and the environment to consider specific factors in design and manufacturing of products						•							
• assess the environmental and economic impact of using conventional and alternative forms of thermal energy								•					
• assess the environmental benefits of technologies that reduce heat loss or transfer								•					
• assess the environmental impacts of processes that change a product through physical or chemical changes													•





Table of Contents

At a Glance™	2
Teacher Assessment Rubric	5
Student Self-Assessment Rubric	6
Introduction	7
Structural Forms	
Teacher Notes	8
Student Activities	9
Forces on Structures	
Teacher Notes	14
Student Activities	15
Form and Stability	
Teacher Notes	20
Student Activities	21
Structure and Function	
Teacher Notes	27
Student Activities	28
Under Construction	
Teacher Notes	35
Student Activities	36
Design and Production	
Teacher Notes	42
Student Activities	44
Temperature Change	
Teacher Notes	48
Student Activities	50
Heat Transfer	
Teacher Notes	56
Student Activities	58
Heat in Our Homes	
Teacher Notes	64
Student Activities	65





Table of Contents

Solutions

Teacher Notes	70
Student Activities	71

Mixtures

Teacher Notes	76
Student Activities	77

Mixing and Separating

Teacher Notes	82
Student Activities	83

Physical or Chemical?

Teacher Notes	91
Student Activities	92





Student's Name: _____ Date: _____

Teacher Assessment Rubric

Success Criteria	Level 1	Level 2	Level 3	Level 4
Knowledge and Understanding Content				
Demonstrate an understanding of the concepts, ideas, terminology definitions, procedures and the safe use of equipment and materials	Demonstrates limited knowledge and understanding of the content	Demonstrates some knowledge and understanding of the content	Demonstrates considerable knowledge and understanding of the content	Demonstrates thorough knowledge and understanding of the content
Thinking Skills and Investigation Process				
Develop hypothesis, formulate questions, select strategies, plan an investigation	Uses planning and critical thinking skills with limited effectiveness	Uses planning and critical thinking skills with some effectiveness	Uses planning and critical thinking skills with considerable effectiveness	Uses planning and critical thinking skills with a high degree of effectiveness
Gather and record data, and make observations, using safety equipment	Uses investigative processing skills with limited effectiveness	Uses investigative processing skills with some effectiveness	Uses investigative processing skills with considerable effectiveness	Uses investigative processing skills with a high degree of effectiveness
Communication				
Organize and communicate ideas and information in oral, visual, and/or written forms	Organizes and communicates ideas and information with limited effectiveness	Organizes and communicates ideas and information with some effectiveness	Organizes and communicates ideas and information with considerable effectiveness	Organizes and communicates ideas and information with a high degree of effectiveness
Use science and technology vocabulary in the communication of ideas and information	Uses vocabulary and terminology with limited effectiveness	Uses vocabulary and terminology with some effectiveness	Uses vocabulary and terminology with considerable effectiveness	Uses vocabulary and terminology with a high degree of effectiveness
Application of Knowledge and Skills to Society and Environment				
Apply knowledge and skills to make connections between science and technology to society and the environment	Makes connections with limited effectiveness	Makes connections with some effectiveness	Makes connections with considerable effectiveness	Makes connections with a high degree of effectiveness
Propose action plans to address problems relating to science and technology, society, and environment	Proposes action plans with limited effectiveness	Proposes action plans with some effectiveness	Proposes action plans with considerable effectiveness	Proposes action plans with a high degree of effectiveness





Name: _____

Date: _____

Student Self-Assessment Rubric

Put a check mark (✓) in the box that best describes you:

	Always	Frequently	Sometimes	Seldom
• I listened to instructions.				
• I was focused and stayed on task.				
• I worked safely.				
• My answers show thought, planning, and good effort.				
• I reported the results of my experiment.				
• I discussed the results of my experiment.				
• I used science and technology vocabulary in my communication.				
• I connected the material to my own life and the real world.				
• I know what I need to improve.				

1. I liked _____

2. I learned _____

3. I want to learn more about _____





Introduction

The activities in this book have two intentions: to teach concepts related to physical science and to provide students the opportunity to apply necessary skills needed for mastery of science and technology curriculum objectives.

Throughout the experiments, the scientific method is used. The scientific method is an investigative process which follows five steps to guide students to discover if evidence supports a hypothesis.

- 1. Consider a question to investigate.**
For each experiment, a question is provided for students to consider. For example, "Does temperature of a solvent affect solubility?"
- 2. Predict what you think will happen.**
A hypothesis is an educated guess about the answer to the question being investigated. For example, "I believe that a heated solvent can dissolve much more solid than a cold solvent". A group discussion is ideal at this point.
- 3. Create a plan or procedure to investigate the hypothesis.**
The plan will include a list of materials and a list of steps to follow. It forms the "experiment."
- 4. Record all the observations of the investigation.**
Results may be recorded in written, table, or diagram format.
- 5. Draw a conclusion.**
Do the results support the hypothesis? Encourage students to share their conclusions with their classmates, or in a large group discussion format.

The experiments in this book fall under thirteen topics that relate to three aspects of physical science: **Structural Form and Function, Heat and Temperature, Chemistry of Pure Substances and Mixtures.** In each section you will find teacher notes designed to provide you guidance with the learning intention, the success criteria, materials needed, a lesson outline, as well as provide some insight on what results to expect when the experiments are conducted. Suggestions for differentiation are also included so that all students can be successful in the learning environment.

Assessment and Evaluation:

Students can complete the Student Self-Assessment Rubric in order to determine their own strengths and areas for improvement. Assessment can be determined by observation of student participation in the investigation process. The classroom teacher can refer to the Teacher Assessment Rubric and complete it for each student to determine if the success criteria outlined in the lesson plan has been achieved. Determining an overall level of success for evaluation purposes can be done by viewing each student's rubric to see what level of achievement predominantly appears throughout the rubric.





Structural Forms

Learning Intention:

Students will learn how to recognize and classify structural forms and the materials used in their construction.

Success Criteria:

- identify structural types as either solid, frame, shell structures
- identify structures as either man-made or natural
- determine the types of materials used to make man-made and natural structures
- determine the properties of the materials
- recognize and describe structural combination forms
- make connections to our environment by identifying structures all around us

Materials Needed:

- a copy of “Structural Forms” Worksheet 1 for each student
- a copy of “What Type Is It?” Worksheet 2 for each student
- a copy of “Properties of Materials ” Worksheet 3 and 4 for each student
- a copy of “Structures All Around” Worksheet 5 for each student
- clipboards (one for each student)
- pencils, pencil crayons, chart paper, markers

Procedure:

1. Discuss with students the meaning of a structure. Lead students in a brainstorming activity/discussion about different structures that can be found in the natural and man-made environments. Record student responses on chart paper. Give students Worksheet 1. After reading and discussing the information on this sheet, refer back to the student responses that were recorded during the brainstorming activity. For each response given, ask students to identify if the structure is man-made or natural, and then identify if it is a solid, frame, or shell type structure. Pose the question, could any of them be a combination of structural types? For example, a house is a combination of a frame and a shell structure. Give students Worksheet 2 to complete.
2. Come back together as a large group and have a discussion about the different materials that man-made structures and those in the environment are made of. Make a list of their responses on chart paper. Discuss the meaning of properties of materials. Refer back to the list of materials on the chart paper, and ask students to name some properties of those materials. Give them Worksheets 3 and 4 to complete.
3. Explain to students that they are going to take a walk around the neighborhood to observe natural and man-made structures. Instruct them to take note of whether the structures they see are natural or man-made, and if they are solid, frame, shell structure types, or a combination. Give each student a clipboard and Worksheet 5 to complete as they walk through the neighborhood.

Differentiation:

Slower learners may benefit by working together with a peer or in a small group with teacher direction to discuss different properties of materials while completing Worksheets 3 and 4.

For enrichment, faster learners could access the internet and research a man-made or natural structure of their choice. Instruct them to find out how exactly this structure was made and what materials were used to make it.

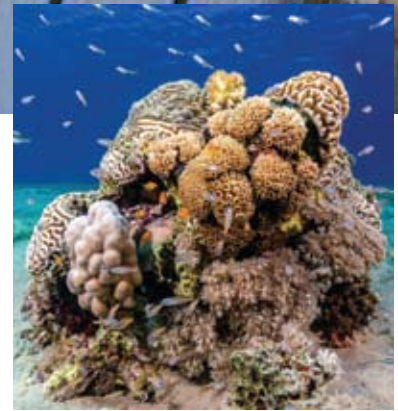




Structural Forms

A structure is a supporting framework that is made to hold a load or to enclose a space. Structures are made from one or more parts. There are many structures all around us. They can be classified by their design into three groups: frame, solid, and shell structures.

Solid structures are sometimes called mass structures. They are usually made from one solid piece of a strong material. A solid structure has little or no space inside and relies on its own weight to resist the forces that act upon it. Generally, the thicker a solid structure is, the stronger it is. An example of a solid structure is a concrete barrier such as a dam, which is able to resist the force of water upon it. We can find solid structures in nature too. Some examples of natural solid structures are mountains or coral reefs.



Frame structures are made up of parts fastened together to create one arrangement. These parts are called its structural components. None of these structural

components are able to singularly support a load, but when components are fastened together, they are able to support and strengthen each other. An example of a frame structure is a bicycle. We can find frame structures in nature too. An example of a natural frame structure is a tree.



In a shell structure, the outside layer of the structure holds the object together. Shell structures have a solid surface and a hollow interior. Strong shells can be made from weak materials, as it is the shape of a shell structure that gives it its strength. An example of a shell structure is the wing of an airplane. We can find shell structures in nature too. An example of a natural shell structure is a snail's shell.





What Type Is It?

Classify the structures in the pictures below.

a) solid, frame, or shell structure? **AND** b) man-made or natural?



a) _____

b) _____



a) _____

b) _____



a) _____

b) _____



a) _____

b) _____



a) _____

b) _____



a) _____

b) _____



a) _____

b) _____



a) _____

b) _____



a) _____

b) _____





Properties of Materials

Structures can be made of different materials. List the kinds of materials that are used to make these man-made structures:



Materials have different properties. Some materials are hard and some are flexible. Hard and flexible are different properties. Use descriptive words to tell about the properties of the man-made structures in the pictures above.





Structures All Around

Take a walk in your neighborhood to observe natural and man-made structures. Draw and label four structures that you see. Also, determine if they are solid, frame, or shell structures.

Circle: natural or man-made?
 solid, frame, or shell?

Circle: natural or man-made?
 solid, frame, or shell?

Circle: natural or man-made?
 solid, frame, or shell?

Circle: natural or man-made?
 solid, frame, or shell?

Do any of the structures that you observed incorporate a combination of structure types? If so, describe the parts of the structure that make up each type.










Physical & Chemical Changes

Solids, liquids, and gases are all matter. When a change to matter occurs, it can be a **physical** or **chemical** change.

Physical change means a change in color, size, or shape. A physical change is a change that can be reversed, or changed back. A change in state such as an ice cube melting to water, is a physical change. By putting the ice cube back into the freezer, the change can be **reversed** and the ice cube can be formed again. Water, ice, and water vapor are all water.

Chemical change means a change in a substance that makes it different from its original matter. A chemical change is a change that can't be reversed, or is very difficult to reverse. A chemical change occurs when a substance changes color, or when a substance gives off bubbles or gas. When we put wood on a campfire, it burns and changes to ashes, gas, soot, and smoke. This is non-reversible as we cannot change the ashes back to wood.

Look at these examples. Tell whether the change is **physical** or **chemical**.

Change	Is it physical or chemical ? Why?
	
	
	
	
	





Reversible or Non-Reversible?

Now you will do some investigating into the types of change. Conduct the following four experiments and decide which changes are reversible and which are non-reversible. Let's get started!

EXPERIMENT #1

Materials needed:

- a beaker of water
- 2 or 3 antacid tablets

What to Do:

1. Drop the antacid tablets into the beaker of water.
2. Record your observations by drawing what happened.
3. Make conclusions about what you observed.

I Observed...	I Conclude that...
	<p><i>(check the boxes that apply)</i></p> <p>The change is: <input type="checkbox"/> reversible <input type="checkbox"/> non-reversible</p> <p>The change is: <input type="checkbox"/> physical <input type="checkbox"/> chemical</p> <p>I know this because _____ _____ _____ _____ _____</p>



**EXPERIMENT #4****Materials needed:**

- lemon juice
- a saucer
- cotton swabs
- blank paper
- lamp

What to Do:

1. Pour some lemon juice into the saucer.
2. Dip a cotton swab into the lemon juice. Use the cotton swab to write a message on the blank paper. Let the lemon juice dry.
3. Hold the paper close to the light. Be sure the paper does not touch the light.
4. Record your observations by drawing what happened.
5. Make conclusions about what you observed.

I Observed...	I Conclude that...
	<p><i>(check the boxes that apply)</i></p> <p>The change is: <input type="checkbox"/> reversible <input type="checkbox"/> non-reversible</p> <p>The change is: <input type="checkbox"/> physical <input type="checkbox"/> chemical</p> <p>I know this because _____ _____ _____ _____ _____</p>

